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APPLICATION NO. FILING DATE FIRST NAMED INVENTOR ATTORNEY DOCKET NO. CONFIRMATION NO. 10/759,261 01/19/2004 William Freeman 15436.121.1.1 6915 22913 7590 12/04/2006 **EXAMINER WORKMAN NYDEGGER** BLACKWELL, GWENDOLYN ANNETTE (F/K/A WORKMAN NYDEGGER & SEELEY) **60 EAST SOUTH TEMPLE** ART UNIT PAPER NUMBER 1000 EAGLE GATE TOWER 1775 SALT LAKE CITY, UT 84111

DATE MAILED: 12/04/2006

Please find below and/or attached an Office communication concerning this application or proceeding.

	Application No.	Applicant(s)
	10/759,261	FREEMAN ET AL.
Office Action Summary	Examiner	Art Unit
	Gwendolyn Blackwell	1775
The MAILING DATE of this commun	nication appears on the cover sheet wi	th the correspondence address
A SHORTENED STATUTORY PERIOD IN WHICHEVER IS LONGER, FROM THE IN Extensions of time may be available under the provision after SIX (6) MONTHS from the mailing date of this come. If NO period for reply is specified above, the maximum is Failure to reply within the set or extended period for reply any reply received by the Office later than three months earned patent term adjustment. See 37 CFR 1.704(b).	MAILING DATE OF THIS COMMUNIO s of 37 CFR 1.136(a). In no event, however, may a re munication. tatutory period will apply and will expire SIX (6) MON y will. by statute, cause the application to become AB	CATION. eply be timely filed THS from the mailing date of this communication. ANDONED (35 U.S.C. § 133).
Status		
,	ed on <u>18 September 2006</u> . 2b) ☐ This action is non-final. In for allowance except for formal matte Fice under Ex parte Quayle, 1935 C.D	
Disposition of Claims		
4)⊠ Claim(s) <u>1-24</u> is/are pending in the 4a) Of the above claim(s) <u>16-24</u> is/a 5)□ Claim(s) is/are allowed. 6)⊠ Claim(s) <u>1-15</u> is/are rejected. 7)□ Claim(s) is/are objected to. 8)□ Claim(s) are subject to restrict the subject the subject the subject to restrict the subject	re withdrawn from consideration.	
Application Papers		•
	2004 is/are: a)⊠ accepted or b)□ olection to the drawing(s) be held in abeyang the correction is required if the drawing(ce. See 37 CFR 1.85(a). (s) is objected to. See 37 CFR 1.121(d).
Priority under 35 U.S.C. § 119		
12) Acknowledgment is made of a claim a) All b) Some * c) None of: 1. Certified copies of the priority 2. Certified copies of the priority 3. Copies of the certified copies application from the Internati	of for foreign priority under 35 U.S.C. § of documents have been received. of documents have been received in A of the priority documents have been onal Bureau (PCT Rule 17.2(a)). on for a list of the certified copies not	pplication No received in this National Stage
Attachment(s) 1) \(\sum \) Notice of References Cited (PTO-892)		Summary (PTO-413)
Notice of Draftsperson's Patent Drawing Review (Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date	PTO-948) Paper No(s	s)/Mail Date nformal Patent Application

DETAILED ACTION

Election/Restrictions

1. Claims 16-24 are withdrawn from further consideration pursuant to 37 CFR 1.142(b) as being drawn to a nonelected method of making, there being no allowable generic or linking claim. Election was made without traverse in the reply filed on September 18, 2006.

Claim Objections

2. Claims 1-7 are objected to because of the following informalities:

Claim 1 refers to an attenuation layer comprising a layer of at least a photopolymerizable monomer mixed with an electrolyte and an electrochromic material. Dependent claim 6 refers to the attenuation layer as comprising an electrochromic layer and an electrolyte layer coupled to said electrochromic layer. From a reading of Applicant's specification (page 6, section 018) the attenuation layer can be one layer or separate layers. Based upon the specification and the language of claims 1 and 6, the attenuation layer was interpreted as separate layers, wherein dependent claim 6 further defines the structure of the attenuation layer of claim 1

However Applicant's arguments (Response, 9/18/2006) indicate that claim 1 should have been an attenuation layer having the monomer, electrolyte, and electrochromic material mixed in one layer. If that is the case, then claim 6 cannot further limit the attenuation layer and make the layer now separate layers claimed as Applicant has indicated in the arguments (Response, pages 8-9) that the attenuation layer is one layer. Clarification is required. To further prosecution, the

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attenuation layer of claim 1 will be construed as being either one layer or multiple layers, which is consistent with the specification, and is further defined as separate layers in claim 6.

Claim Rejections - 35 USC § 102

3. The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

4. Claims 1-3, 5-10, 12, and 14-15 are rejected under 35 U.S.C. 102(b) as being anticipated by United States Patent no. 5,668,663, Varaprasad et al.

Regarding claims 1 and 8

Varaprasad et al disclose an electrochromic device, such as a window (column 1, lines 6-11) comprised of first and second substrates positioned in a spaced apart relationship being substantially parallel. First and second conductive electrodes are positioned on the inner surfaces (the surfaces that face each other) of the first and second substrates, (columns 13-14, lines 45-41). Between the conductive electrodes, an electrochromic material layer and an electrolyte material layer (the combined electrochromic layer/electrolyte layer form Applicant's attenuation layer) are formed, (column 7, lines 53-64). The electrolyte material is comprised of redox reaction promotors and alkali ions and/or protons wherein one of the alkali ions may be lithium methacrylate (photopolymerizable element), (columns 8-10, lines 56-58), meeting the limitations of claims 1 and 8.

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Regarding claims 2-3, 5-7, 9-10, 12, and 14-15

The substrates can be formed of glass, (column 13, lines 26-41), meeting the limitations of claims 2 and 9.

The electrodes and glass substrates are transparent and transmissive in part in the visible portion of the electromagnetic spectrum, (column 15, lines 6-65), meeting the limitations of claims 3 and 10.

As light passes through the electrolyte layer, a portion of the electromagnetic spectrum is absorbed (attenuated), (columns 11-12, lines 61-8), meeting the limitations of claim 5.

Between the conductive electrodes, an electrochromic material layer and an electrolyte material layer (the combined electrochromic layer/electrolyte layer form Applicant's attenuation layer) are formed, (column 7, lines 53-64), meeting the limitations of claim 6.

The electrochromic/electrolyte layers are activated by an applied potential between the conductive electrode coatings by any source of an alternating current or a direct current (voltage), (column 23, lines 39-49), meeting the limitations of claim 7.

The electrolyte material is comprised of redox reaction promotors and alkali ions and/or protons wherein one of the alkali ions may be lithium methacrylate (photopolymerizable element) which form either a liquid or solid solution, (columns 8-10, lines 56-58), meeting the limitations of claims 12 and 14-15.

5. Claims 1-4 and 7-13 are rejected under 35 U.S.C. 102(b) as being anticipated by United States Patent no. 6,193,378, Tonar et al.

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Regarding claims 1, 4, 8, and 11-13

Tonar et al disclose an electrochromic device that can be a window, (column 11, lines 47-50 and column 12, lines 53-57). The device is comprised of first and second substrates with a first electrode on the inner surface of the first substrate and a second electrode on the inner surface of the second substrates wherein the two substrates are in a spaced apart relationship with an electrochromic element formed between the two electrodes, (column 3, lines 38-67).

Regarding claims 2-3, 7, and 9-10

The substrates are made of glass, (column 10, lines 17-24), meeting the limitations of claims 2 and 9.

The conductive electrodes are transparent, which would allow for the transmission of at least a portion of visible light, (column 3, lines 57-65), meeting the limitations of claims 3 and 10.

The reflectivity of electrochromic element is activated through the use of an applied voltage, (column 3, lines 37-41), meeting the limitations of claim 7.

Response to Arguments

- 6. Applicant's arguments filed September 18, 2006 have been fully considered but they are not persuasive.
- 7. Applicant contends that US 5,668,663, (Varaprasad), does not teach the structure of the attenuation layer as claimed in claims 1 and 8 as the attenuation layer is comprised a layer of photopolymerizable monomer mixed with an electrolyte and an electrochromic element.

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With regards to claim 1, this is not persuasive as claim 1 has been given the broadest reasonable interpretation based upon the specification and those claims which depend therefrom. Claim 6, requires that the attenuation layer is comprised of separate layers with the layers being an electrolyte layer and an electrochromic layer. With such an interpretation, it would stand to reason that the electrolyte layer is comprise of the electrolyte and the photopolymerizable monomer, as claim 1 indicates that the photopolymerizable monomer is mixed with an electrolyte. The electrolyte layer of Varaprasad contains a photopolymerizable monomer which could also be stated as a photopolymerizable monomer mixed with an electrolyte.

With regards to claim 8, the argument is not persuasive as it is not commensurate with the language of claim 8. There is no limitation in claim 8 that the photopolymerizable monomer is mixed with an electrolyte and an electrochromic material.

8. Applicant contends that US 6,193,378, (Tonar), teaches an electrochromic material that is not claimed by Applicant.

This is not persuasive as the entire rejection of Tonar set forth the elements of the structure of claims 1 and 8. Applicant has not established that Tonar does not teach those elements. Applicant is arguing the definition of a term outside the proper context of the entire rejection under Tonar. The electrochromic solution, which should have been the electrochromic material, is:

The electrochromic element is comprised of an electrolyte and an electrochromic medium, (column 4, lines 46-67). The example demonstrates that the electrochromic medium also contains polymethylmethacrylate (photopolymerizable monomer), (column 10, lines 32-37), meeting the limitations of claims 1, 4, 8, and 11-13.

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The electrochromic elements is a mixture of the electrochromic medium, electrolyte, and photopolymerizable monomer, which are the same the elements of Applicant's attenuation layer.

9. For the reasons set forth above, the rejections of claims 1 and 8 are maintained. As Applicant only argued against claims 1 and 8, the rejection of dependent claims 2-7 and 8-15 are maintained.

Conclusion

10. THIS ACTION IS MADE FINAL. Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Gwendolyn Blackwell whose telephone number is (571) 272-1533. The examiner can normally be reached on Monday - Thursday; 6:30 am - 5:00 pm.

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If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Jennifer McNeil can be reached on (571) 272-1540. The fax phone number for the

organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent

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Gwendolyn Blackwell Examiner

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Call

JENNIFER MCNEIL SUPERVISORY PATENT EXAMINER

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